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韩群,秦亚玲,李德峰.细菌 Rieske 非血红素铁环羟化酶在多环芳烃降解中的研究进展.生物工程学报,2021,37(10): 3439-3458.

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摘 要: 多环芳烃是一种常见的持久性有机污染物,因具有致癌、致突变等毒性而被广泛关注。其微生物 降解过程通常由羟化起始,随后脱氢、开环、一步步去除支链,最终进入三羧酸循环。Rieske 非血红素铁 环羟化酶 (Rieske-type non-heme iron aromatic ring-hydroxylating oxygenases, RHOs,又称 aromatic ring-hydroxylating dioxygenases) 或细胞色素 P450 氧化酶负责将羟基加成到多环芳烃环上,将疏水性的多 环芳烃转化为亲水性的衍生物,这一过程是多环芳烃降解转化的起始步骤,也是关键步骤和限速步骤之一。 文中主要介绍 RHOs 的分布、底物特异性、底物识别机制以及研究 RHOs 与多环芳烃的一些技术和方法等, 并对 RHOs 在环境修复技术中的潜在应用进行了展望。

关键词: Rieske 非血红素铁环羟化酶,多环芳烃,底物范围,底物识别,环境修复

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Advances in bacterial Rieske non-heme iron ring-hydroxylating dioxygenases that initiate polycyclic aromatic hydrocarbons degradation

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Abstract: Polycyclic aromatic hydrocarbons (PAHs) are a class of persistent organic pollutants, which have received widespread attentions due to their carcinogenic and mutagenic toxicity. The microbial degradation of PAHs are usually started from the hydroxylation, followed by dehydrogenation, ring cleavage and step-by-step removal of branched chains, and finally mineralized by the tricarboxylic acid cycle. Rieske type non-heme iron aromatic ring-hydroxylating dioxygenases (RHOs) or cytochrome P450 oxidases are responsible for the conversion of hydrophobic PAHs into hydrophilic derivatives by the ring hydroxylation is the first step of PAHs degradation and also one of the rate-limiting steps. Here, we review the distribution, substrate specificity, and substrate recognition mechanisms of RHOs, along with some techniques and methods used for the research of RHOs and PAHs.

Keywords: Rieske non-heme iron ring-hydroxylation dioxygenases, polycyclic aromatic hydrocarbons (PAHs), substrate range, substrate recognition, environmental remediation

1 多环芳烃和 RHOs 双加氧酶

1.1 多环芳烃

多环芳烃 (Polycyclic aromatic hydrocarbons, PAHs) 是一种常见的持久性有机污染物,其分子 结构中含有两个及以上苯环, 疏水性强, 水溶性 低,能够稳定地存在于环境中,不易被降解^[1], 因其具有致癌、致突变等毒性而被广泛关注^[2]。 多环芳烃根据分子量的高低,分为由萘、菲、蒽 等 2 个或 3 个稠环组成的低分子量多环芳烃和由 荧蒽、芘、苯并[a] 蒽和苯并[a] 芘等 4 个或 4 个以 上环组成的高分子量多环芳烃^[3]。在国内外,多 环芳烃都是非常受关注的环境污染物。1995年, 美国环保署就将 16 种多环芳烃列为优先控制的 污染物^[4],我国《土壤环境质量建设用地土壤污 染风险管控标准 (试行)》(GB36600-2018)^[5]等国 家标准将萘、苯并[a] 蒽等 8 种 PAHs 列入环境污 染风险筛选和管控范围 (图 1)。目前, 多环芳烃 有物理、化学和生物等降解方式^[6]。其中,生物 降解是一种自然降解方式,占环境中多环芳烃降

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解的 40%-60%^[7],是去除或者降低环境中多环芳 烃污染的一种重要方式^[8]。当前已经从多环芳烃 污染的环境中分离到多种以多环芳烃为唯一碳源 的菌株,如能降解萘、菲的 Diaphorobacter sp.、 寡养单胞菌属 Stenotrophomonas sp.^[9-10], 以及能 够降解高分子量多环芳烃荧蒽、芘的红球菌属 Rhodococcus sp.、假单胞菌属 Pseudomonas sp.、 鞘氨醇盒菌属 Sphingopyxis sp.、芽孢杆菌属 Bacillus sp.^[1,11]等。国内很多实验室也分离获得对 萘^[12]、菲、荧蒽^[13]、苯并[a]芘^[14]等多种多环芳烃 具有高效降解作用的菌株,比如恶臭假单胞菌 Pseudomonas putida B6-2 对 13 种多环芳烃及7种 二噁英都具有一定的降解能力^[15],而油菜假单胞 菌 Pseudomonas brassicacearum MPDS 对多环芳 烃和杂环衍生物都具有高效的降解能力^[16]。另 外,我国科学家在多环芳烃污染的来源、时间空 间分布^[17]、健康风险评估^[18-19]、生物修复^[20]、代 谢途径 (如下游水杨酸途径^[21]、龙胆酸代谢途 径^[22]等) 以及相关酶 (如环羟基化酶^[23]等) 分子



图 1 美国环保署优先控制的 16 种多环芳烃 (黑色框为国内管控的 8 种多环芳烃) Fig. 1 Sixteen PAHs regulated by EPA of US (black boxed are the eight PAHs regulated in China).

机制研究等方面做了大量的突出工作。目前低 分子量多环芳烃及其下游代谢途径的研究已经 比较清楚,尤其是萘、菲、蒽、芴等的降解途 径^[24]。而由于高分子量多环芳烃的水溶性更差、 代谢途径更长,也更复杂多样化^[25],对于细菌 中高分子量多环芳烃降解途径的研究不甚透 彻,在已经报道的高分子量多环芳烃降解菌株 中的具体代谢途径及其中所涉及的酶的鉴定及 其性质等还需进一步探究。可以预见在较长一 段时间内,针对高致畸性高危害性高分子量多 环芳烃的降解机制和代谢途径研究将是多环芳 烃研究的重点。

1.2 多环芳烃降解关键酶——RHOs

多环芳烃的降解起始于羟化,然后脱氢、开 环、一步步去除支链,最终进入三羧酸循环 (Tricarboxylic acid cycle, TCA cycle)被微生物所 利用。微生物降解多环芳烃的第一步反应即羟化 反应,这一步在细菌中研究较多的酶是 Rieske 非 血红素铁环羟基化酶 (Rieske-type non-heme iron aromatic ring-hydroxylating dioxygenases, RHOs), 在真菌中研究较多的是细胞色素 P450 单加氧酶 (Cytochrome P450 monooxygenase, CYP450)。 RHOs 能催化多种氧化反应,包括碳-碳键的氧化 裂解、单羟基化和二羟基化反应^[26]。这种酶是唯

一已知的催化邻位顺式二醇立体选择性一步形成 的酶^[27],能在一个酶促步骤中将两个羟基引入到 多环芳烃上形成顺式二氢二醇 (图 2)[28]。顺式二 氢二醇经过脱氢酶脱氢后形成相应的二醇、在内 二醇或外二醇双加氧酶的作用下开环,内二醇双 加氧酶作用于两个羟基之间的 C-C 键, 而外二醇 双加氧酶切割与两个羟基相邻的 C-C 键底物^[29]。 例如, PhdF 就是一种 I 型外二醇双加氧酶, 是邻 位氧螯合超家族的成员,可作用于 3,4-二羟基菲 等含有双羟基的苯环上,发挥开环的功能^[30]。 CYP450在人类、植物、微生物中是广泛存在的, 而且其底物范围非常广泛,能够催化 C-羟基化及 杂原子氧合、释放和环氧化物的形成等^[31]。 CYP450参与了PAHs生物降解过程的初始氧化步 骤、催化多环芳烃的不同位置形成一种或多种单 羟基化产物^[32],或者与环氧化物水解酶协同作用 催化形成反式-二氢二醇 (图 3)^[33-34]。CYP450 主 要在真菌中研究的较多,如白腐真菌^[35]、黄孢 原毛平革菌,且能够作用于四环、五环乃至六 环的苯并[ghi] 花等高分子量的多环芳烃^[36]、细 菌中少见报道。而在 Rhodococcus sp. P14^[37]和 Mycobacterium vanbaalenii PYR-1^[38]中都提到存 在 RHOs 与 CYP450 共同降解多环芳烃的现象, 这些菌能同时催化高分子量 PAHs 产生顺式-二氢 二醇和反式-二氢二醇。而且研究者证明在多株 Mycobacterium vanbaalenii 菌中都有 RHOs 与 CYP450 基因共同存在的情况^[38]。前面提到,内 二醇以及外二醇双加氧酶只能识别两个羟基中间 或者与两个羟基相邻的 C-C 键, 而 CYP450 作用 多环芳烃后一部分仅产生一个羟基形成酚,因 而推测在 CYP450 羟化后,可能由 CYP450 再进 行一次羟化,或者由 RHOs 对 CYP450 单羟基化



图 2 RHOs 起始催化萘的降解途径

Fig. 2 Degradation pathway of naphthalene initiated by RHOs.



图 3 CYP450 催化菲

Fig. 3 Catalytic activity of CYP450 on phenanthrene.

的产物再次羟基化,最终形成开环酶能够作用的 二醇的结构。而且早期也有文章提到 RHOs 能够 催化单加氧反应^[39],如甲苯单加氧酶可能催化苯 酚生成邻苯二酚^[40-41], 萘双加氧酶也被证实能参 与多种苄基单羟基化反应^[42]。也有证据表明萘双 加氧酶能够催化荧蒽和芴单羟基化分布形成 8-羟 基荧菌和 9-芴醇^[43],以及其他的研究中 RHOs 催 化的单羟基化产物的出现^[44]。RHOs 作为高分子 量多环芳烃降解的两类起始酶之一,一旦确定特 定多环芳烃可以被某一 RHOs 氧化后,即有可能 根据基因组分布、代谢物分析等推测相关代谢中 间物和中间涂径,具有"提纲挈领"的意义。但是 RHOs 的生化研究过程中存在诸多需要被克服的 困难。从技术上来说,首先,多环芳烃特别是高 分子量多环芳烃溶解性差[24],在体外酶活系统构 建中与酶的结合能力差,需要有机溶剂溶解,但 有机溶剂又可能对酶活性造成损伤^[43]。RHOs体 外酶活系统中所需的电子传递体中的铁氧还蛋 白还原酶 (Ferredoxin reductase, FdR) 难以纯化 表达^[43],给体外酶活测定带来一定的困难,选 择外源的电子传递体不能保证双加氧酶的酶活 效率,在确定合适的电子传递体前,酶的活性 难以用传统的 NADH 的消耗来表征;另外,由 于高分子量的多环芳烃环数多, RHOs 作用的位 置难以确定,产生一种甚至多种产物^[45],而对于 产物的鉴定多采用的是气相色谱-质谱联用技术 (GC-MS),因为产物缺少标准品或者标准品价格 昂贵,使得研究方法受限。而在生信分析中,RHOs 的底物范围非常广泛,以萘双加氧酶 (Naphthalene dioxygenase, NDO) 为例^[42], 它可以以多种物质 为底物,这也为酶的底物确定带来了一定的困难。

综上,细菌 RHOs 在多环芳烃的降解途径中 发挥着非常重要的作用,也存在着一定的研究难 度和迫切之处,下面我们将对与多环芳烃降解相 关的 RHOs 进行详细的论述。

2 Rieske 非血红素铁环羟基化双加氧酶 (RHOs) 的来源、组成

2.1 RHOs 的菌株来源分布

目前对于 RHOs 的研究集中在 Rhodococcus sp.、Pseudomonas sp.、罗尔斯通氏菌属 Ralstonia sp.、海杆菌属 Marinobacter sp.、从毛单胞菌属 Comamonas sp.、潘多拉菌属 Pandoraea sp.等菌 株中[46]。表 1 统计了 1988 年至今能够使用如 Naphthalene dioxygenase, Biphenyl dioxygenases 等特定关键词查询到的在文献中发表的已经与多 环芳烃降解相关的 RHOs 基因,少有发现新型的 RHOs, 进行表征的都与早期发现的 RHOs 具有高 度同源性。在多环芳烃降解菌的筛选及研究过程 中还发现 RHOs 可能在不同菌属中存在水平转移 的现象。早期有观点认为 RHOs 可能起源于革兰 氏阴性细菌,随后转移到革兰氏阳性细菌中^[47]。 水平转移的现象被研究者们多次提到,如:利用 简并 PCR 引物从 Marinobacter 中扩增萘双加氧酶 的大亚基, 经系统发育树分析, 该萘双加氧酶与 Pseudomonas 和伯克氏菌属 Burkholderia 来源的 萘双加氧酶相似,而在其基因簇附近发现了 tnpA1 基因的同源基因,可能编码噬菌体 λ 型转 座酶,这可能解释了多环芳烃降解基因在这些 细菌谱系中发生的水平转移^[48]。同样的,以来自 Pseudomonas spp.的 2Fe-2S 还原酶基因设计引物, 发现食酸菌属 Acidovorax 中有与 Pseudomonas spp.相同的萘双加氧酶的相关基因,但是属于同 一物种的菌株则没有^[49]。研究推测这种基因水平 转移的发生在微生物群落适应污染物中发挥了重 要作用[50]。

2.2 Rieske 非血红素铁环羟基化酶 (RHOs) 的组成

Rieske 非血红素铁环羟基化酶 (RHOs) 是 一种多组分酶系统,由末端双加氧酶、铁氧还蛋白 (Ferredoxin, Fd)、铁氧还蛋白还原酶 (Ferredoxin

表 1 RHOs 来源及其作用的底物				
Table	Origins and substrates of RHOs			

GenBank	Strains	RHOs	Substrates	Year	References
CP054128	Pseudomonas sp. MPDS	Naphthalene	Naphthalene, fluorene,	2021	[16]
		dioxygenase	dibenzofuran and		
			dibenzothiophene		
MH560349	Pseudomonas fluorescens	Naphthalene	Phenanthrene	2020	[51]
	AH-40	dioxygenase			
KJ461700	Pseudomonas aeruginosa	Naphthalene	Naphthalene	2020	[52]
	N6P6	dioxygenase			
MULN0000000	Pseudomonas veronii	Naphthalene	Naphthalene	2019	[53]
	strainVI4T1	dioxygenase			
JN613334	Rhodococcus	Naphthalene	Phenanthrene	2019	[54]
	wratislaviensis strain 9	dioxygenase			
AAD28100	Rhodococcus sp. strain	Naphthalene	Naphthalene	2019	[55]
	NCIMB12038	dioxygenase			
CP006254	Geobacillus sp. JF8	Naphthalene	Naphthalene	2019	[56]
		dioxygenase			
NO	Sphingobium yanoikuyae	Biphenyl	Phenazine	2017	[57]
	B1	2,3-dioxygenase			
KC771235	Pseudomonas aeruginosa	Biphenyl dioxygenase	Biphenyl	2016	[58]
	JP-11				
JN235141	<i>Rhodococcus</i> sp. ustb-1	Naphthalene	Pyrene	2015	[59]
0.4/070	D 4	dioxygenase		2015	5.603
Q46372	Pandoraea pnomenusa	Biphenyl dioxygenases	3-hydroxy-4,4'-dichlorob	2015	[60]
	B356		iphenyl,		
			3,3 -dinydroxy-4,4 -		
			chlorobiphenyl, flavone,		
VN102520	TT 1/ 1	A	isoflavone, and flavanone	2014	[(1]
KM102520	Uncultured	Aromatic	Biphenyl, naphthalene,	2014	[61]
	gammaproteobacterium	ring-nydroxylating	pnenanthrene, pyrene,		
WM102522	The evolutions of	oxygenases (pnd20/19)	Diskerel seekthelese	2014	[(1]
KW1102322		Aromatic ring hydroxylating	Biphenyi, naphthalene	2014	[01]
	gammaproteobacterium	oruganasas (hph20/28)			
KM102523	Unculturad	Aromatic	Rinhanyl nanhthalana	2014	[61]
KW1102323	aammaprotoobactorium	ring hydroxylating	phononthrono	2014	[01]
	gammaproreobacierium	oxygenases (nah33/32)	phenantmene		
IN655512	Comamonas sp. MO	Nanhthalene	Indole and most indole	2013	[62]
31(055512	comunionus sp. mg	dioxygenase	derivatives	2015	[02]
NO	Martelella sp AD-3	Naphthalene	Anthracene	2012	[63]
110	munerena sp. mb s	dioxygenase		2012	[00]
P37333	Pseudomonas strain LB400	A variant biphenvl	Dibenzofuran	2012	[64]
		dioxygenase			[*.]
GO184726	Burkholderia sp. C3	<i>Nag</i> -like dioxygenases	Naphthalene.	2011	[65]
	1	, <u>,</u>	dibenzothiophene		[]
GQ184727	Burkholderia sp. C3	Phn-like dioxygenases	Naphthalene,	2011	[65]
	I		phenanthrene,		
			dibenzothiophene		
Q53122	Rhodococcus jostii RHA1	Biphenyl and	Polybrominated diphenyl	2011	[66]
		ethylbenzene	ethers		
		dioxygenases			

					(续表 1)
GenBank	Strains	RHOs	Substrates	Year	References
SRA028415	Sediment metagenome	Novel aromatic ring hydroxylating dioxygenases	Biphenyls	2011	[67]
DQ846881	Rhodococcus opacus R7	Naphthalene dioxygenase	Naphthalene	2010	[68]
Q3C1D5	Comamonas sp. strain E6	Terephthalate	Terephthalate	2008	[69]
Q53122	Rhodococcus jostii RHA1	Biphenyl and ethylbenzene dioxygenases	Styrene and benzene	2008	[70]
Q53122	Rhodococcus jostii RHA1	Biphenyl 2,3-dioxygenase	Biphenyl/polychlorinated- biphenyl	2007	[71]
HE577117	Uncultured bacterium	Biphenyl dioxygenases	Biphenyls	2007	[72]
EF152282	Sphingobium yanoikuyae B1	Biphenyl/naphthalene dioxygenase	Biphenyl, naphthalene, and phenanthrene, toluene, m- and p-xylene	2007	[73]
AF295032	<i>Marinobacter</i> strain NCE312	Naphthalene dioxygenase	Naphthalene and 2-methylnaphthalene	2006	[48]
Q46372	Pandoraea pnomenusa B356	Biphenyl dioxygenases	2-hydroxy-3- chlorobiphenyl, 2-hydroxy-5- chlorobiphenyl and 2-hydroxy-3,5- dichlorobiphenyl	2004	[74]
P37333	<i>Pseudomonas</i> strain LB400	Biphenyl dioxygenases	2-hydroxy-3- chlorobiphenyl, 2-hydroxy-5- chlorobiphenyl and 2-hydroxy-3,5- dichlorobiphenyl, 2,2'-dichlorobiphenyl	2004	[74-75]
O52382	Ralstonia sp. strain U2	Naphthalene dioxygenase	Naphthalene	2002	[76]
AF061751	<i>Burkholderia</i> sp. strain RP007	<i>Phn</i> -like dioxygenases	Naphthalene, phenanthrene	1999	[77]
Q46372	Comamonas testosteroni strain B-356	Biphenyl dioxygenases	Biphenyl/chlorobiphenyl dioxygenase	1996	[78]
A5W4F2	Pseudomonas putida F1	Toluene dioxygenase	Benzene and toluene	1994	[79]
Q52438	<i>Pseudomonas</i> sp. strain KKS102	Biphenyl 2,3-dioxygenase	Biphenyl and polychlorinated biphenyls	1994	[55]
Q07944	Pseudomonas putida ML2	Benzene dioxygenase	Benzene	1993	[80]
Q52028	Pseudomonas pseudoalcaligenes KF707	Biphenyl dioxygenase	Biphenyls and polychlorinated biphenyls	1992	[81]
P37333	Pseudomonas strain LB400	Biphenyl dioxygenases	Polychlorinated-biphenyl	1992	[82]
P0A110	Pseudomonas putida strain NCIB9816	Naphthalene dioxygenase	Indole	1988	[83]

reductases, FdR)^[84]三部分组成(图 4)。末端加 氧酶一般含有大亚基和小亚基,构成 α₃β₃ 异六 聚体结构,如 Sphingobium yanoikuyae B1 中的 联苯双加氧酶^[85];或者是只含有大亚基,构成 α₃ 同三聚体,如咔唑的末端加氧酶成分 (PDB ID: 4NBD)^[86]。末端双加氧酶的 α 亚基包含一个 Rieske [2Fe-2S]簇和一个单核铁活性中心。Fd 含 有一个 Rieske 型的[2Fe-2S]簇^[87], 而 FdR 是由 3个结构域组成: FAD 结合结构域、NADH 结合 结构域和C末端结构域^[88],有些菌中的Fd和FdR 可能融合形成一个蛋白,如食醚红球菌 Rhodococcus aetherivorans IcdP1^[89] (GenBank 登 录号: CP011341) 中的 RHOs 加氧酶组分附近分 布着一些潜在的天然融合 NAD/FAD 或 FMN/NAD 结合域和[2Fe-2S]簇的蛋白,可能是 为 RHOs 的加氧酶组分发挥电子传递作用,这种 融合蛋白发挥电子传递作用早在 1992 年就有报 道,如黄素蛋白与[2Fe-2S]簇的天然融合^[90]。 RHOs 的 3 个组分行使功能时,首先是 FdR 从 NAD(P)H 中释放电子,并将电子转移到 Fd 中,

Fd 将电子转移到加氧酶^[91],加氧酶的 Rieske [2Fe-2S]簇从电子转移组分接受电子后,将电子转移到活性位点的单核铁,激活分子氧,从而攻击 底物^[26]。

3 RHOs 双加氧酶的底物范围

萘双加氧酶 (Naphthalene dioxygenase, NDO) 是与多环芳烃降解相关的 Rieske 非血红素铁环羟 化酶中研究较多的一类。萘双加氧酶有多种类型, nag 类基因是从 Pseudomonas putida strain G7 首 次 发 现 的 能 够 降 解 萘 的 基 因 , phn 是 从 Burkholderia sp. RP007 中发现的能以萘、菲和蒽 为唯一碳源的萘降解基因^[65]。NDO 的底物范围非 常广,可作用于萘之外的多种物质,具体可参考 表 1 所统计的部分 RHOs 的底物。例如,荧光假 单胞菌 Pseudomonas fluorescens AH-4、Rhodococcus wratislaviensis strain 9 的 NDO 能氧化菲^[51,54]。萘 双加氧酶还能作用于吲哚产生靛蓝或靛玉红^[92]; 鞘氨醇单胞菌 Sphingomonas CHY-1 中的 NDO 能 羟化荧蒽、苯并[a]蒽、菌、苯并[a]芘、芘,其中,





Fig. 4 Structures of three RHOs components. (A) Dioxygenase component (PDB ID: 2CKF). (B) Ferredoxin reductase (PDB ID: 1F3P). (C) Ferredoxin (PDB ID: 2QPZ).

苯并[a]芘的分子量虽然比芘更大,但是 NDO 作 用于苯并[a]芘时的酶活更高,可能是在与 NDO 接触时, 苯并[a]芘多出来的一个环更能靠近狭窄 的酶活性中心^[93]。另外 RHOs 中还有一类与联苯 型多环芳烃相关的联苯双加氧酶 (Biphenyl dioxygenase, BPDOs) 的研究也相对较多。BPDOs 作用的底物多与多氯联苯及其衍生物相关,如 Pandoraea pnomenusa B356的 BPDOs 能够催化联 苯,后续发现它可将 5-氯-2-羟基联苯转化为 5-氯-2-羟基苯甲酸酯,还可参与对双对位氯取代 联苯类似物 (3-羟基-4,4'-二氯联苯和 3,3'-二羟 基-4,4'-氯联苯)的代谢^[60]。Pseudomonas strain LB400 也可降解联苯,同 B-356 BPDOs 一样具有 可以催化 2-羟基-3-氯联苯、2-羟基-5-氯联苯和 2-羟基-3,5-二氯联苯的能力^[74]。另外 LB400 还具 有催化 2,2'-二氯联苯的能力^[75]。除此之外,有研 究者还发现 Rhodococcus jostii RHA1 中的联苯和 乙苯双加氧酶在联苯、乙苯等底物存在下能够转 化多溴二苯醚 (PBDEs)^[66]。

4 基于结构生物学的底物识别机制研究

RHOs 可以催化多种反应,与许多不同底物 形成复合物的结构表明,活性位点中底物的取向 不仅控制区域特异性和立体特异性,而且还控制 催化的反应类型^[94]。在 RHOs 结合底物后,没有 发现明显的侧链重排^[91],底物与单核铁活性中心 接近的两个碳被羟基化,这解释了由 RHOs 产生 的顺式二氢二醇的极端区域和立体选择性^[91]。底 物必须与 RHOs 充分相互作用,以防止催化时底 物发生运动,催化活性口袋内部若存在多余空间 可能会影响 RHOs 在发生催化作用时底物在口袋 内部的活动^[95]。

表2所示是PDB数据库中部分已解析结构的 RHOs。RHOs利用一个单核非血红素铁中心来进 行催化,此过程消耗了2个电子、2个质子和 1 个氧气分子,产生了顺式二氢二醇。NDO 的结构表明活性位点口袋内的氨基酸大部分是疏水性的,这为芳香族底物的结合提供了合适的环境^[96]。 *Pseudomonas* sp. NCIB 9816-4 的 NDO 结构是第 一个被解析的 RHOs^[97]。它的第 352 位突变影响 了萘的立体选择性,206 位和 295 位影响了联苯 和菲的区域选择性,在同时突变 206/352 或者 206/295/352 的情况下,NDO 催化菲形成菲顺 式-9,10-二氢二醇作为主要产物^[98],表明催化位点 附近的关键残基影响 RHOs 的底物识别。

相对于催化萘的 RHOs (图 5A) 来说,结合 高分子量多环芳烃的 RHOs 的疏水性底物的结合 口袋明显更大,如 Sphingomonas CHY-1 的 RHOs, CHY-1 的 RHOs 是比较典型的能够羟基化五环 PAH 的酶, 它的催化腔的中心区域主要由 Phe350、Phe404 和 Leu356 的侧链形成, 形成相 当均匀的梯形腔,从而影响酶的位置特异性^[99]。 另外, 天然和联苯结合形式的 BPDO-OB1 结构与 萘 1,2-双加氧酶相似,但活性位点入口也明显大 于萘 1.2-双加氧酶的入口,活性位点残基的差异 也允许高分子量多环芳烃菌和苯并[a]芘的结 合^[100]。除此之外, RHOs 活性位点入口存在的柔 性环也被认为是一种可以扩大底物特异性范围的 特征,这个环可以在需要时提供重塑活性位点所 需的结构灵活性,有利于不同配体的调节^[61]。 CHY-1 的 RHOs 的柔性环 L1 上的 Leu223 和 L2 上的 Ile260 有助于催化位点对高分子量多环芳烃 的选择性 (图 5B)。同样,在 BphAE_{LB400}的 α 亚 基上的残基也有一段覆盖活性位点的环,在底物 结合时,该片段发生了明显的位移^[95]。在鞘氨醇 菌属 Sphingobium sp. FB3 中 RHOs 的加氧酶组 分,223 位是 Phe,比能羟基化五环 PAH 的 Sphingomonas CHY-1 的 RHOs 的 Leu 大得多, 突 变该位点后 FB3-RHOs 降解苯并[a]芘速率加 快^[14]。这也说明关键残基的改变就可能改变酶的

底物特异性,再如 BphAEs 从 280 到 283 位于催 化腔的入口,而 283 位的 Ser 被 Met 取代后,改 变了催化位点内底物的方向,从而改变了其羟基 化位点,增强了它对于多种多氯联苯的特异性。 也正说明 283 位的突变对 BPDO 的底物特异性有显著影响^[101]; BPDO 中与底物识别和区域特异性相关的位点被突变后也改变了 BPDO 在苯环上的作用位点^[102]。

表 2 已解析结构的 RHOs

Table 2RHOs with resolved structure

PDB ID	Strains	RHOs	Substrates	Year	References
4HJL, 4HKV, 4HM0-8	Pseudomonas sp. C18	Naphthalene 1,2-dioxygenase	1-chloronaphthalene, benzamide, indole-3-acetate, thioanisole, styrene, indene, phenetole, indan, ethylbenzene, ethylphenylsulfide, 1-indanone	2012	[103]
2YFI	ParaBurkholderia xenovorans LB400	Biphenyl dioxygenase variant Rr41	Dibenzofuran	2011	[104]
2XR8	ParaBurkholderia xenovorans LB400	Biphenyl dioxygenase	Many polychlorinated biphenyls	2010	[105]
3EN1	Pseudomonas putida	Toluene 2,3-dioxygenase	Toluene	2010	[88]
3GZY	Comamonas testosteroni sp. strain B-356	Biphenyl dioxygenase	Biphenyl and polychlorinated biphenyls	2009	[106]
2CKF	<i>Sphingomonas</i> sp. CHY-1	PAH-hydroxylating dioxygenase	Fluoranthene, benz[a]anthracene, benzo[a]pyrene	2007	[99]
2HMM	Pseudomonas sp.	Naphthalene 1,2-dioxygenase	Anthracene	2006	[91]
2HML	Pseudomonas sp.	Naphthalene 1,2-dioxygenase mutant	Phenanthrene	2006	[91]
2HMK	Pseudomonas sp.	Naphthalene 1,2-dioxygenase	Phenanthrene	2006	[91]
2GBX	Sphingomonas yanoikuyae B1	Biphenyl 2,3-dioxygenase	Biphenyl	2006	[100]
2DE7	Janthinobacterium	Carbazole 1,9a-dioxygenase	Carbazole	2006	[107]
2B24	<i>Rhodococcus</i> sp. NCIMB 12038	Naphthalene 1,2-dioxygenase	Indole	2005	[108]
2BMR	<i>Comamonas</i> sp. JS765	Nitrobenzene dioxygenase	3-nitrotoluene	2005	[109]
1WQL	Pseudomonas fluorescens Ip01	Cumene dioxygenase	Cumene or toluene	2004	[110]
1WW9	Janthinobacterium sp. J3	Carbazole 1,9a-dioxygenase	Carbazole	2005	[111]
1UUV	Pseudomonas putida	Naphthalene 1,2-dioxygenase	Nitric oxide and indole	2004	[112]
1ULI	<i>Rhodococcus jostii</i> RHA1	Biphenyl dioxygenase	Biphenyl	2003	[113]
107G	Pseudomonas putida	Naphthalene 1,2-dioxygenase	Naphthalene	2002	[114]
1EG9	Pseudomonas putida	Naphthalene 1.2-dioxygenase	Indole	2000	[115]
1NDO	Pseudomonas putida	Napthalene 1,2-dioxygenase	Naphthalene	1998	[116]



图 5 RHOs 底物结合口袋

Fig. 5 The ligand binding pocket of RHOs. (A) The ligand binding pocket of NDO (PDB ID: 107G) from *Pseudomonas* sp. NCIB 9816-4. The naphthalene molecule was shown in yellow and some important residue in green. (B) Comparison with RHOs from *Sphingomonas* CHY-1 (Blue, PDB ID: 2CKF) and from *Pseudomonas* sp. NCIB 9816-4 (green). Residues Leu223 and Ile260 with smaller amino acid side chains indicated the possibility of that *Sphingomonas* CHY-1 RHOs bind PAHs molecules with larger molecular weight.

5 一些针对 RHOs 和多环芳烃的研究方法

5.1 环境中 RHOs 的筛选

用唯一碳源的无机盐培养基从污染环境中富 集分离多环芳烃降解菌株是普遍采用的方式。通 过污染物压力筛选能产生与多环芳烃降解相关的 微群落,进而研究其中的 RHOs^[117]。环境中还存 在着非常多的不可培养微生物发挥降解作用,因 此研究人员利用分子生物学的方法来筛选和检测 环境中的 RHOs, 如建立 RHOs 的加氧酶组分的 α-亚基基因序列数据库,以保守的 Rieske 中心的 核酸序列,设计高度特异性的引物以进行 RHOs 的丰度或多样性等的研究[118]。一系列基于保守区 域引物的技术被应用,如聚合酶链式反应 (Polymerase chain reaction, PCR)^[119]、利用双加 氧酶探针进行 Southern 杂交实验^[120]、多重 PCR 技术等来鉴定和检测芳香族双加氧酶基因,以及 利用定量PCR扩增技术量化环境样品中的芳香族 分解代谢基因^[121]。检测到环境中存在的 RHOs 后,通过构建宏基因组文库,如以 Fosmid、 Cosmid、BAC 为载体的文库构建,可以获取到相 关 RHOs 的全长序列^[122],能够对其功能进行研 究及表征,扩展对于未培养微生物中 PAH 降解 基因的了解,更能够扩大对于 RHOs 的认知,找 到更多功能强大的 RHOs 来用于场地或相关环境 的修复。

5.2 大肠杆菌异源表达-体内活性测定

RHOs 功能表征的一个研究思路是将 RHOs 的双加氧酶组分、电子传递组分分别克隆到不同 质粒上,转到同一宿主中进行诱导表达,若要检 测这些基因是否正确表达产生了有降解活性的 RHOs,则可以采用靛蓝筛选法^[123]、高效液相色 谱 (HPLC) 或气相色谱 (GC) 检测代谢物,使用 氧气电极检测实时活性^[124]。

其中靛蓝筛选法是最为方便快捷的方法, 萘 双加氧酶型 RHOs 的特征是吲哚向靛蓝转化, 添 加不高于 4 mmol/L 的吲哚 (高于 4 mmol/L 对细 胞有毒性)能使得菌株中的活性萘双加氧酶 (NDO) 生成靛蓝^[125],大肠杆菌细胞可以利用色 氨酸酶从色氨酸合成吲哚,所以含有 RHOs 的克 隆不需要添加吲哚即可验证在大肠杆菌中的活 性^[126],从而作为筛选 NDO 的表征。而 HPLC 或 GC-MS (质谱) 联用是最常用的多环芳烃降解产 物的检测和鉴定方法^[3,93,127]。但该方法操作较为 复杂,要对代谢产物经过反复萃取-旋蒸浓缩-干 燥除水-重溶-过滤等步骤,不适合高通量的筛选, 但是能较为精确地定量。基于高碘酸盐荧光检测 RHOs 产生的顺式二醇代谢物的方法是一种新的 筛洗方法,其采用高碘酸钠氧化顺式二醇代谢物 转化为相应的二醛,用以产生可检测的分析物^[128]。 此方法与传统方法相比优势在于能够高通量地筛 选双加氧酶或者有羟基化多环芳烃能力的突变体 双加氧酶。

5.3 重组蛋白的体外活性与底物结合

RHOs 的体外酶活测定需要完整的酶组分才 能进行,因而需要将加氧酶组分和 Fd、FdR 电子 传递组分都进行纯化表达,得到纯度较高的蛋白 质,通过在 340 nm 处 NADH 的消耗来测定酶活 性^[61],或者利用氧电极测定耗氧速率^[129]。此外, 差示扫描荧光法 (Differential scanning fluorimetry, DSF) 是一种基于蛋白质的热稳定性原理高通量 筛选蛋白质与配体相互作用的方法,在其他蛋白 质的配体筛选中已经被广泛使用。此种方法结合 上文提到的高通量的荧光检测方法来检测 RHOs 与多环芳烃的混合物孵育后的产物中是否有顺式 二醇代谢物,不失为一种简单的检测方法,为 RHOs 的底物筛选提供了一种高通量的方式。

而重组蛋白测活的难点在于 FdR 的表达存在 问题,大部分重组蛋白表达为包涵体的形式,而 这个问题不能通过改变宿主菌株或在诱导过程中 降低温度来解决^[43],睾丸酮丛毛单胞菌 *Comamonas testosterone* B-356 联苯双加氧酶的 FdR 组分表达时也遇到了相同的问题,最后使用 QIAGEN His-Binding Kit 纯化得到能用于实验的 蛋白量^[130]。从类产碱假单胞菌 Pseudomonas pseudoalcaligenes KF707 中克隆表达联苯双加氧 酶的 FdR 组分时,用了一个含有分子伴侣 groELS 基因的质粒 pKY206 成功表达了可溶性的 FdR 蛋 白^[131]。Sphingomonas sp. strain CHY-1 的 FdR 组 分克隆到 pET15b 载体上转化到大肠杆菌 BL21(DE3) 中进行体内活性测定时,发现比没有 Fd、FdR 组分的菌株对菲的降解率增加了 35 倍, 说明 FdR 在大肠杆菌中正确合成并发挥作用^[132], 但后期纯化时只能获得少量的该酶,也不能进行 大量的表达,因而更换了其他菌株中更为稳定的 FdR 进行实验,并不影响电子传递作用^[43]。以 Pseudomonas putida F1 中的甲苯双加氧酶进行了 实验验证, 在更换 Fd 的情况下, 不更换 FdR 能 获得原来活性的 56%, 而更换合适的 FdR 也能使 得活性达到原来的 38%^[84]。因而当 FdR 不能正确 表达时,可以使用分子伴侣辅助 FdR 表达,也可 尝试更换其他来源的 FdR 电子传递组分进行酶活 测定。

5.4 分子模拟

通过同源建模、分子对接模拟分析多环芳烃 与双加氧酶活性中心的相互作用机理从而解释实 验现象是一种比较普遍的方法^[23]。研究 RHOs 底 物特异性的方法已经不仅仅局限于实验结果,从 计算机的分子模拟中可获得底物和酶的结构信 息,并可以解释生化实验中的一些现象,为实验 提供参考性的帮助,而且通过对接分析预测的大 多数底物是可以通过实验验证的^[126]。目前的同源 建模可以采用 Geno3D、SWISS-MODEL、PHYRE、 MODELLER 和 Schrödinger。建模的结果可通过 PROCHECK^[59]获取到 Ramachandran 图来评分, 或者通过 VADAR 软件评估单个残基和整个蛋白 质的关键结构参数来评估模型的质量,结构错误 可通过 WHAT-CHECK 进行分析^[126],氨基酸序列 与其 3D 原子模型之间的相容性通过 VERIFY-3D 进行评估^[59,129]。MOLE 2.0 识别从溶剂区域通向 活性部位的可能通道^[133], POCASA 1.0 可预测催 化口袋的区域^[126],分子对接的软件可使用 AutoDock Vina^[134]或者 Sybyl7.3 中的 Surflex 模 块,用于探索 NDO 活性位点与配体之间的相互 作用机理, Surflex 可以有效地减少假阳性结果的 百分比^[59]。分子模拟在 RHOs 的研究中是一个非 常好的工具,它能够在获得蛋白质与多环芳烃复 合物结构之前推测蛋白与底物的相互作用。对 RHOs 进行设计改造以产生更有效的酶活,适应 环境实际需求来进行环境修复方面,分子模拟是 非常重要的^[135]。

6 潜在应用与展望

Rieske 非血红素铁环羟基化双加氧酶在多环 芳烃降解中发挥着重要的作用,它是唯一能催化 多环芳烃形成顺式二氢二醇的酶,与多环芳烃污 染修复有着紧密关联。其一,根据 RHOs 的保守 性,能对环境中存在的多环芳烃降解菌株进行定 量及筛选,结合宏基因组学相关技术^[122],能够获 取到更多不可培养微生物的基因信息,不但丰富 对于不可培养微生物的认识,更能够获取到新的 高效降解多环芳烃的基因。其二, RHOs 可以作 为一种多环芳烃污染场地的指示物,针对 RHOs 保守区域制备相应的引物即可监测受污染场所的 芳烃降解菌株的种群丰度[118]。其三,在长期污染 的环境中, RHOs 是可以在菌群中水平转移的, 从而使得微生物能够适应这种极端恶劣环境^[50]。 人工模拟自然条件下的水平转移,可获得含有快 速进化的高效 RHOs 的微生物,将其应用到污染 环境中,能够更好地适应环境以降解污染物。其 四,伴随着合成生物学的发展,通过分子模拟 来设计和改造 RHOs,结合定向突变技术^[101],能 够获得更多有益的 RHOs 的突变体, 扩宽 RHOs 的底物催化范围,一种底物范围广、高效而且稳 定的酶制剂或人工微生物对多环芳烃污染治理有

着潜在的应用价值。其五,多环芳烃诸多的降解基 因簇也被用于构建基因工程微生物 (Genetically engineered microorganisms, GEMs),如 nah-like 基因簇等,GEMs 可被用来清理污染场地,提高 污染物的去除能力及范围^[136]。目前,已经有一些 RHOs 或利用基因工程技术改造过的 RHOs 被申 请专利,未来有潜力被用来处理受到有毒或致癌 芳香族化合物污染的土壤、湿地和污水^[137-139]。 对于多环芳烃污染场地的生物修复来说,RHOs 有着广阔的应用前景。

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